

<b>Linear Algebra I</b>			
<b>Registration code</b>	0061211	<b>Credits</b>	2.0
<b>Course Category</b>	Sciences Basic	<b>Class room</b>	A11
<b>Term(Semester)/Day/Period</b>	I (First Year, First Semester) / Mon / 2 (10:30~12:00)		
<b>Instructor</b>	DARPÖ Erik		
<b>Target Schools (Programs)</b>	Le(J) • La(S) • Ec(S) • Sc(P • C • B) • En(P • C • Au) • Ag(B)		
<p>● <b>Aim of the course</b>            Linearity one of the most fundamental concepts for the handling of quantities in current natural science. Indispensable in quantum mechanics and relativity, its use has spread across all branches of natural science and beyond. Linear algebra, developed in the nineteenth century, is the mathematical theory of linearity. The first half of this one-year course focuses on the techniques for manipulating systems of linear equations and their application to analytic geometry (in arbitrary dimensions). Emphasis is placed on the ability to think abstractly.</p> <p>● <b>Course Prerequisites</b>            Ability to deal with systems of linear equations and to do elementary geometry in the plane can help.</p> <p>● <b>Outline of lectures</b>            I) Geometric setting            Points and vectors in <math>\mathbb{R}^n</math>, located vectors in <math>\mathbb{R}^n</math>, norm and scalar product in <math>\mathbb{R}^n</math>, parametric representation of lines, planes and hyperplanes.            II) Systems of linear equations            Systems of linear equations, row operations and Gaussian elimination.            III) Vector subspaces            Definition, linear combinations, linear independence of vectors, dimension of a subspace.            IV) Linear maps and matrices            Linear maps, matrices associated with linear maps, inverse, elementary matrices and Gaussian elimination for matrices. Kernel and image (range) of linear maps, rank of linear maps and matrices, composition of linear maps.</p> <p>● <b>Course Evaluation Methods</b>            There will be two main, written exams: midterm (35%) and final (45%). Additionally, there will be homework assignments (10%) and quizzes (10%). The final grade will be determined by the total amount of points obtained according to the following scale: S: 90-100, A: 80-89, B: 70-79, C: 60-69, F: 0-59.</p> <p>● <b>Notice for Students</b>            The Reference Book is available in the Main library and in the Science library (enough copies in total for all students).</p> <p>It is <b>strongly</b> recommended to also follow the course Mathematics Tutorial I b.</p>			
<b>Textbook</b>	---		
<b>Reference Book</b>	<i>Linear Algebra with Applications</i> , fourth edition, Otto Bretscher, Edition: Pearson		

<b>Fundamentals of Biology I</b>													
<b>Registration code</b>	0061311	<b>Credits</b>	2.0										
<b>Course Category</b>	Sciences Basic	<b>Classroom</b>	A21										
<b>Term(Semester)/Day/Period</b>	I (First Year, First Semester) / Mon / 3 (13:00~14:30)												
<b>Instructor</b>	CARTAGENA Joyce Abad												
<b>Contact</b>	Office: Rm. B508A, Building B, Graduate School of Bioagricultural Sciences Phone: 052-789-5209 / Mobile: 09082038167 (for emergency only) Email: joyce@agr.nagoya-u.ac.jp												
<b>Office hours:</b>	Wednesdays 3:00-6:00 pm Fridays 3:00-6:00 pm *Email or call me for appointments outside office hours.												
<b>Target Schools (Programs)</b>	Sc (P · C · B) · En (P · C · Au) · Ag (B)												
<p>● <b>Aim of the course</b> The aim of this course is to introduce the key concepts of basic biology. In order to understand life, we have to understand the molecular processes that occur in and around cells, from energy procurement to reproduction and gene expression. Through such processes, living things go through evolution giving rise to biological diversity.</p> <p>● <b>Course Prerequisites</b> None</p> <p>● <b>Course Content</b></p> <table style="width: 100%; border: none;"> <tr> <td style="width: 50%; vertical-align: top;"> <p>I. THE LIFE OF THE CELL</p> <ol style="list-style-type: none"> <li>1. Biology: Exploring Life</li> <li>2. The Chemical Basis of Life</li> <li>3. The Molecules of Cells</li> <li>4. A Tour of the Cell</li> <li>5. The Working Cell</li> <li>6. How Cells Harvest Chemical Energy</li> <li>7. Photosynthesis: Using Light to Make Food</li> </ol> <p>II. CELLULAR REPRODUCTION AND GENETICS</p> <ol style="list-style-type: none"> <li>8. The Cellular Basis of Reproduction and Inheritance</li> <li>9. Patterns of Inheritance</li> <li>10. Molecular Biology of the Gene</li> <li>11. How Genes Are Controlled</li> <li>12. DNA Technology and Genomics</li> </ol> </td> <td style="width: 50%; vertical-align: top;"> <p>III. CONCEPTS OF EVOLUTION</p> <ol style="list-style-type: none"> <li>13. The Origin of Species</li> <li>14. Tracing Evolutionary History</li> <li>15. How Populations Evolve</li> </ol> <p>IV. THE EVOLUTION OF BIOLOGICAL DIVERSITY</p> <ol style="list-style-type: none"> <li>16. Microbial Life: Prokaryotes and Protists</li> <li>17. The Evolution of Plant and Fungal Diversity</li> <li>18. The Evolution of Invertebrate Diversity</li> <li>19. The Evolution of Vertebrate Diversity</li> </ol> </td> </tr> </table> <p>● <b>Course Evaluation Methods</b></p> <table style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="text-align: center; border-top: 1px solid black; border-bottom: 1px solid black;">Activity</th> <th style="text-align: center; border-top: 1px solid black; border-bottom: 1px solid black;">Percentage of final grade</th> </tr> </thead> <tbody> <tr> <td style="border-bottom: 1px solid black;">In-class participation (includes recitation/class discussions)</td> <td style="text-align: center; border-bottom: 1px solid black;">25</td> </tr> <tr> <td style="border-bottom: 1px solid black;">Home works and Quizzes</td> <td style="text-align: center; border-bottom: 1px solid black;">25</td> </tr> <tr> <td style="border-bottom: 1px solid black;">Examinations</td> <td style="text-align: center; border-bottom: 1px solid black;">50</td> </tr> </tbody> </table>				<p>I. THE LIFE OF THE CELL</p> <ol style="list-style-type: none"> <li>1. Biology: Exploring Life</li> <li>2. The Chemical Basis of Life</li> <li>3. The Molecules of Cells</li> <li>4. A Tour of the Cell</li> <li>5. The Working Cell</li> <li>6. How Cells Harvest Chemical Energy</li> <li>7. Photosynthesis: Using Light to Make Food</li> </ol> <p>II. CELLULAR REPRODUCTION AND GENETICS</p> <ol style="list-style-type: none"> <li>8. The Cellular Basis of Reproduction and Inheritance</li> <li>9. Patterns of Inheritance</li> <li>10. Molecular Biology of the Gene</li> <li>11. How Genes Are Controlled</li> <li>12. DNA Technology and Genomics</li> </ol>	<p>III. CONCEPTS OF EVOLUTION</p> <ol style="list-style-type: none"> <li>13. The Origin of Species</li> <li>14. Tracing Evolutionary History</li> <li>15. How Populations Evolve</li> </ol> <p>IV. THE EVOLUTION OF BIOLOGICAL DIVERSITY</p> <ol style="list-style-type: none"> <li>16. Microbial Life: Prokaryotes and Protists</li> <li>17. The Evolution of Plant and Fungal Diversity</li> <li>18. The Evolution of Invertebrate Diversity</li> <li>19. The Evolution of Vertebrate Diversity</li> </ol>	Activity	Percentage of final grade	In-class participation (includes recitation/class discussions)	25	Home works and Quizzes	25	Examinations	50
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**●Notice for Students**

1. Course webpage

NUCT (Nagoya University Collaboration and Course Tools) is an online system that will be used for this course. Home works will be accessible through this page as well as extra learning materials that I will be uploading. Moreover, this webpage will also be used as a venue for us to communicate.

<https://ct.nagoya-u.ac.jp/portal>

2. Attendance

If you cannot attend class, you should contact me as soon as possible either by email or phone. Attendance will not be graded but missing class would mean missing possible points from recitation.

3. Make-up exam

Make-up exams may be given on condition that the student can provide acceptable reasons for his/her absence.

4. Personal electronics policy

Personal electronic devices should not be visible or audible during class time.

5. Academic honesty and original work

Cheating and copying (including plagiarism) will not be tolerated in this class.

6. Course Withdrawal

Students who wish to withdraw from the course will have to submit a duly accomplished Course Withdrawal Request by November 21, 2016.

<b>Textbook</b>	Campbell Biology Concepts and Connections 8/e 2015 (Pearson New International Edition) *or older edition Authors: J. Reece, M. Taylor, E. Simon, J. Dickey
<b>Reference Book</b>	---

<b>Fundamentals of Earth Science I</b>			
<b>Registration code</b>	0061411	<b>Credits</b>	2.0
<b>Course Category</b>	Sciences Basic	<b>Classroom</b>	A11
<b>Term(Semester)/Day/Period</b>	I (First Year, First Semester) / Mon / 4 (14:45~16:15)		
<b>Instructor</b>	HUMBLET Marc Andre		
<b>Contact</b>	Office: Graduate School of Environmental Studies Department of Earth and Planetary Sciences E516 Phone: 052-789-3037 Email: humblet.marc@f.mbox.nagoya-u.ac.jp		
<b>Target Schools (Programs)</b>	Sc(P · C · B) · En(P · C · Au) · Ag(B)		
<p>● <b>Aim of the course</b></p> <p>The study of planet Earth embraces a wide range of topics, from the formation of rocks to the evolution of life. In this course, we will talk about plate tectonics, the fundamental theory underlying all geological processes which have shaped the environment in which we live and continue to modify the landscape, from the slow and progressive uplift of mountain chains to violent earthquakes and volcanic eruptions. The students will learn how the Earth recycles matter and how minerals and rocks form and are transformed. One chapter of the course is dedicated to the issue of time, central in earth science, and tackles the question of how the age of rocks and geological events can be determined. We will then take a step back and look at Earth's 4.5 billion year history to see how the Earth's geography has changed and how life has evolved.</p> <p>Besides providing a basic and up-to-date knowledge of the essential concepts of earth sciences, the aim of this course is to stimulate the interest and curiosity of the students for the study of planet Earth and provoke questions, comments, and discussions about issues related to earth sciences.</p> <p>● <b>Content of the course</b></p> <ol style="list-style-type: none"> <li>1. Earth Sciences: an introduction</li> <li>2. The solar system</li> <li>3. Plate tectonics</li> <li>4. Minerals: rock's elementary building blocks</li> <li>5. Rocks and rock cycle I: igneous rocks</li> <li>6. Rocks and rock cycle II: sedimentary rocks</li> <li>7. Rocks and rock cycle III: metamorphic rocks</li> <li>8. The age of rocks</li> <li>9. Earth history I: paleogeography</li> <li>10. Earth history II: origin and evolution of life</li> </ol> <p>● <b>Practical classes</b></p> <p>The students will examine hand-size rock samples and rock thin sections chosen to illustrate the different rock types and geological structures seen during the course. In addition, the students will also participate in a one-day field trip to examine the geology of Mizunami area (Gifu Prefecture), examine Miocene fossils and sediments (20-15 million years old), and learn how geologists collect data in the field.</p> <p>● <b>Grading</b></p> <p>Two quizzes (multiple choice): 20% (10% each) Mid-term exam: 40% Final exam: 40%</p> <p>Students will be graded following the five-step S-A-B-C-F grade evaluation system. S: 90-100%, A: 80-89%, B: 70-79%, C: 60-69%, F: 59-0%</p> <p>A student will be given an "Absent" grade if he or she submits a Course Withdrawal Request by the 15<sup>th</sup> of November. This deadline does not apply to students who drop the class part-way through for an exceptional reason (e.g. illness, accident).</p> <p>● <b>Notice for Students</b></p> <p>Handouts of lecture notes and slides will be distributed during the class. Students can refer to the reference books indicated if they wish to have complementary information about the subjects covered by the course. The books are available at the science library.</p>			
<b>Textbook</b>	---		
<b>Reference Books</b>	John Grotzinger, Understanding Earth 6/e (ISBN:9781429240031 or 9781429219518) Diane Carlson, Physical Geology International Edition (ISBN:9780071221849)		

<b>Fundamentals of Physics I</b>											
<b>Registration code</b>	0062211	<b>Credits</b>	2.0								
<b>Course Category</b>	Sciences Basic	<b>Classroom</b>	Tue-C25 / Thu-A21								
<b>Term(Semester)/Day/Period</b>	I (First Year, First Semester) / Tue, Thu / 2 (10:30~12:00)										
<b>Instructor</b>	FOONG See Kit										
<b>Contact</b>	Email: skfoong@eken.phys.nagoya-u.ac.jp										
<b>Target Schools (Programs)</b>	Sc(P · C · B) · En(P · C · Au) · Ag(B)										
<p>● <b>Aim of the course</b></p> <p>Fundamentals of Physics I (FP I) is the first of four lecture courses (FP I-IV) designed to cover the basic classical physics that provide a firm foundation in learning science and engineering. This course introduces the concepts and laws of classical mechanics. Further topics in mechanics will be covered in FP II.</p> <p>● <b>Course Prerequisites</b></p> <p>Students without a good background in high school physics and basic calculus are advised to review those materials as soon as possible and would be expected to have to spend more time and effort for the course, which must be considered when deciding your course load. Students are expected to participate actively in class activities throughout the course.</p> <p>● <b>Course Content</b></p> <p>The topics include kinematics, vectors, force and motion, energy, work and momentum, and are based on the following chapters in the textbook:</p> <table style="width: 100%; border: none;"> <tr> <td style="width: 50%;">Chapter 2: Motion Along a Straight Line</td> <td style="width: 50%;">Chapter 6: Force and Motion II</td> </tr> <tr> <td>Chapter 3: Vector</td> <td>Chapter 7: Kinetic Energy and Work</td> </tr> <tr> <td>Chapter 4: Motion in Two and Three Dimensions</td> <td>Chapter 8: Potential Energy and Conservation of Energy</td> </tr> <tr> <td>Chapter 5: Force and Motion I</td> <td>Chapter 9: Center of Mass and Linear Momentum</td> </tr> </table> <p>Problem solving involving the applications of the concepts and laws will be discussed in lectures, but the companion course - Fundamental Physics Tutorial Ia - is designed to develop students' problem solving skills.</p> <p>● <b>Course Evaluation Methods</b></p> <p>Class attendance: 5%      Continuous Assessment: 15%      Mid-Term Exam: 30%      Final Exam: 50%</p> <p>Class attendance is required. Absentee must give a valid reason supported by documents. The ABSENT grade will be assigned when a) your attendance is below 75%, OR b) you are absent without valid reason from the Mid-Term Exam or Final Exam., OR c) you wish to receive the ABSENT grade (In this case, you must see the instructor immediately after the Final Exam.)</p> <p>● <b>Notice for Students</b></p> <p>Concurrent registration of Fundamental Physics Tutorial Ia is required. Students who receive the ABSENT grade for the course are not qualified to take the Repeat Exam.</p>				Chapter 2: Motion Along a Straight Line	Chapter 6: Force and Motion II	Chapter 3: Vector	Chapter 7: Kinetic Energy and Work	Chapter 4: Motion in Two and Three Dimensions	Chapter 8: Potential Energy and Conservation of Energy	Chapter 5: Force and Motion I	Chapter 9: Center of Mass and Linear Momentum
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<b>Textbook</b>	Principles of Physics, Extended 9th or 10th Edition International Student Version with WileyPLUS Set (John Wiley & Sons), ISBN (9th Ed): 9780470561584; ISBN(10th Ed): 9781118230749										
<b>Reference Books</b>	<ol style="list-style-type: none"> <li>1. The physics of everyday phenomena, a conceptual introduction to physics by W.T.Griffith and J.W. Brossing (McGraw-Hill, 8<sup>th</sup> Ed 2014)</li> <li>2. Force + Motion, an illustrated guide to Newton's laws by Jason Zimba (John Hopkins University Press, 2009)</li> <li>3. Fundamentals of Physics 8/E Student Solutions Manual by J. R. Christman (Wiley, 2008)</li> <li>4. Feynman Lectures On Physics (Vol.1) by Richard P. Feynman (Pearson P T R)</li> </ol>										

<b>Fundamentals of Physics II</b>			
<b>Registration code</b>		0064211	<b>Credits</b> 2.0
<b>Course Category</b>		Sciences Basic	<b>Classroom</b> Tue-C25 / Thu-A21
<b>Term(Semester)/Day/Period</b>		I (First Year, First Semester) / Tue, Thu / 2 (10:30~12:00)	
<b>Instructor</b>		TAMA Florence Muriel	GELLOZ Bernard Jacques
<b>Contact</b>	<b>Office</b>	ITbM Building / 3 <sup>rd</sup> floor	Eng.Bld3, North wing 431
	<b>Phone</b>	---	052-789-4202
	<b>Email</b>	florence.tama@nagoya-u.jp	gelloz@nuap.nagoya-u.ac.jp
<b>Office Hours</b>		TBA	
<b>Target Schools (Programs)</b>		Sc(P · C · B) · En(P · C · Au) · Ag(B)	
<b>Fundamentals of Physics II (Classical Mechanics II and Thermal Physics)</b>			
<p>● <b>Aim of the course</b></p> <p>Physics is at the foundation of science and engineering. This is the second of a series of four courses that cover the fundamentals of physics. The first 2/3 of this course covers further topics in mechanics: equilibrium and elasticity, gravitation, oscillations and the remaining 1/3 of the course introduces thermal physics. Besides learning to solve problems within each topic, students will also learn to solve problems that cut across these topics.</p>			
<p>● <b>Course Prerequisites</b></p> <p>-To take Fundamentals of Physics II, you must also enroll in Fundamentals of Physics I. (You cannot study Fundamentals of Physics II without taking Fundamentals of Physics I first.) -Note that this course commences after Fundamentals of Physics I; nevertheless, you must register for it during the normal registration period in the first few weeks of semester. -Concurrent registration for Fundamental Physics Tutorial is required. -Students are expected to participate actively in class activities throughout the course. Students without a good background in high school physics and basic calculus are expected to have to spend more time in this course, and are advised to take this into consideration when deciding their course load.</p>			
<p>● <b>Course Content</b></p> <p>Chapter 10: Rotation  Chapter 11: Rolling, Torque, and Angular Momentum  Chapter 12: Equilibrium  Chapter 13: Gravitation  Chapter 15: Oscillations  Chapter 18: Temperature, Heat, and the First Law of Thermodynamics  Chapter 19: The Kinetic Theory of Gases  Chapter 20: Entropy and the Second Law of Thermodynamics</p>			
<p>● <b>Course Evaluation Methods</b></p> <p>Class attendance is required. Absentees must give a valid reason (e.g. doctor's certificate).  The "Absent" grade is reserved for students who withdraw just after the final exam. After that day, a letter grade will be awarded based on marks earned from all assessment during the semester.  Class attendance: 5%; Assignments: 15%; Intermediate tests: 40%; Final Exam: 40%</p>			
<p>● <b>Notice for Students</b></p> <p>Students gain a functional understanding of introductory mechanics and thermal physics. They are able to solve problems that may cut across the topics, and are able to appreciate the physics underlying their studies in other science and engineering disciplines. They are prepared for the next course in the series: Fundamentals of Physics III.</p>			
<b>Related courses:</b> Calculus I, Calculus II, Linear Algebra I, Linear Algebra II, Fundamentals of Physics I, III & IV.			
<b>Textbook</b>	Fundamentals of Physics Extended 10th Edition International Student Version with WileyPLUS Set (John Wiley & Sons, 2010 ISBN: 9781118230749)		
<b>Reference Book</b>	Feynman Lectures in Physics (Vol.1) by Richard Feynman (Pearson P T R)		

<b>Fundamentals of Chemistry I</b>																			
<b>Registration code</b>	0063311	<b>Credits</b>	2.0																
<b>Course Category</b>	Sciences Basic	<b>Classroom</b>	A31																
<b>Term (Semester)/Day/Period</b>	I (First Year, First Semester) / Wed / 3 (13:00 – 14:30)																		
<b>Instructor</b>	BUTKO Peter																		
<b>Contact</b>	Office: SA Building-318-1 (Science & Agriculture) Phone: 052-789-2480 Email: pbutko@chem.nagoya-u.ac.jp																		
<b>Target Schools (Programs)</b>	Sc(P · C · B) · En(P · C · Au) · Ag(B)																		
<p>● <b>Course Purpose</b> The purpose of this course is to grasp what chemistry is all about and to learn important principles and facts in chemistry. The course begins with atomic structure, proceeds next to bonding and molecules, and further to bulk physical properties of substances.</p>																			
<p>● <b>Course Contents</b></p> <table style="width: 100%; border: none;"> <tr> <td style="width: 50%;">1 Chemistry: Matter and Measurement (Ch. 1)</td> <td style="width: 50%;">9 Thermochemistry: Chemical Energy (Ch. 8)</td> </tr> <tr> <td>2 Atoms, Molecules and Ions (Ch. 2)</td> <td>10 Pre-exam Review &amp; <b>EXAM 2 (Chs. 5 – 8)</b></td> </tr> <tr> <td>3 Mass Relationships in Chemical Reactions (Ch. 3)</td> <td>11 Gases: Their Properties and Behavior (Ch. 9)</td> </tr> <tr> <td>4 Reactions in Aqueous Solutions (Ch. 4)</td> <td>12 Liquids, Solids, and Phase Changes (Ch. 10)</td> </tr> <tr> <td>5 Pre-exam Review &amp; <b>EXAM 1 (Chs. 1 – 4)</b></td> <td>13 Solutions and Their Properties (Ch. 11)</td> </tr> <tr> <td>6 Periodicity &amp; the Electronic Structure of Atoms (Ch. 5)</td> <td>14 Pre-final Review</td> </tr> <tr> <td>7 Ionic Bonds &amp; Some Main-Group Chemistry (Ch. 6)</td> <td><b>15 FINAL EXAM (Chs. 1 – 11)</b></td> </tr> <tr> <td>8 Covalent Bonds and Molecular Structure (Ch. 7)</td> <td></td> </tr> </table>				1 Chemistry: Matter and Measurement (Ch. 1)	9 Thermochemistry: Chemical Energy (Ch. 8)	2 Atoms, Molecules and Ions (Ch. 2)	10 Pre-exam Review & <b>EXAM 2 (Chs. 5 – 8)</b>	3 Mass Relationships in Chemical Reactions (Ch. 3)	11 Gases: Their Properties and Behavior (Ch. 9)	4 Reactions in Aqueous Solutions (Ch. 4)	12 Liquids, Solids, and Phase Changes (Ch. 10)	5 Pre-exam Review & <b>EXAM 1 (Chs. 1 – 4)</b>	13 Solutions and Their Properties (Ch. 11)	6 Periodicity & the Electronic Structure of Atoms (Ch. 5)	14 Pre-final Review	7 Ionic Bonds & Some Main-Group Chemistry (Ch. 6)	<b>15 FINAL EXAM (Chs. 1 – 11)</b>	8 Covalent Bonds and Molecular Structure (Ch. 7)	
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<p>● <b>Grading</b> Two Exams: 100 points each. Final Exam (comprehensive): 200. Homework: 50. TOTAL: 450. Grade "S": 100-90% (405 or more points), "A": 89-80% (404 - 360 pts), "B": 79-70% (359 - 315 pts), "C": 69-60% (314 - 270 pts), "F": 59-0% (fewer than 270 pts).</p>																			
<p>● <b>Course Withdrawal</b> <b>Yes.</b> The last day to withdraw without academic penalty is the last class day in November.</p>		<p>● <b>Criteria for “Absent” &amp; “Fail” Grades</b> The “Absent” grade is reserved for students that withdraw by last class day in November. After that day, a letter grade will be awarded based on grades earned from all assignments during the semester.</p>																	
<p>● <b>Prerequisite</b> None</p>		<p>● <b>Related Courses</b> ---</p>																	
<b>Textbook</b>	Chemistry (J. McMurry and R.C. Fay), 7th Ed. (Global Edition, bundled with Mastering Chemistry) Pearson, 2016																		
<b>Reference Book</b>	Chemistry (J. McMurry and R.C. Fay) 6th Ed. International Edition, bundled with Mastering Chemistry (without Etext) ¥9,950																		
<b>Remarks</b>	<p>It is essential to sit in each exam during the scheduled class time. <b>There will be NO make-up exam.</b> In the event of a missed exam due to a serious illness, accident or family emergency, compelling <b>written</b> documentation of the reason for the absence will be required. If the reason is accepted, the final grade will be calculated from the appropriately weighted average from the rest of the exams. If the reason will be deemed insufficient, the absence will be unexcused, and zero points will be awarded for the missed exam. <b>WARNING: Missing more than one exam (it does not matter whether excused or not) means automatically failing the course.</b></p> <p><b>Attendance</b> is necessary for successful completion of this course. No points will be awarded for attending lectures, but attendance may be taken. Sleeping in the lecture hall will be actively discouraged.</p> <p><b>Homework</b> is crucial for mastering new material and developing skills in applying concepts. Weekly homework will be either on paper or electronic. Homework is due at the beginning of class on the due date. <b>A general guideline says an average of 2 to 3 hours of study time per week is necessary for each 1 credit hour.</b></p> <p><b>Exams</b> focus on problem solving, and exam questions will be similar to the homework problems. Exam grades will be posted in the Gradebook on the Course website before next class period.</p> <p><b>Cell phones</b> must be turned off during lecture.</p>																		

<b>Pre-college Mathematics * Optional subject</b>			
<b>Registration code</b>	0063411	<b>Credits</b>	2.0
<b>Course Category</b>	Sciences Basic	<b>Classroom</b>	C10
<b>Term(Semester)/Day/Period</b>	I (First Year, First Semester) / Wed / 4 (14:45~16:15)		
<b>Instructor</b>	RICHARD Serge Charles		
<b>Contact</b>	Office: Room 237, Science Building A Email: richard@math.nagoya-u.ac.jp Web site : <a href="http://www.math.nagoya-u.ac.jp/~richard/fall2016.html">http://www.math.nagoya-u.ac.jp/~richard/fall2016.html</a>		
<b>Target Schools (Programs)</b>	Le(J) · La(S) · Ec(S) · Sc(P · C · B) · En(P · C · Au) · Ag(B)		
<p>● <b>Aim of the course</b> This course is a companion course to Calculus I. It aims to help students with little or no precalculus knowledge to master the basic calculus material in preparation for the more advanced course of Calculus I.</p> <p>● <b>Course Prerequisites</b> No prerequisites.</p> <p>● <b>Course Content</b> The content of this course will depend on the initial level in mathematics of the students attending it. It will mainly consist in a review of high school mathematics and in an additional help for students attending the course Calculus I.</p> <p>● <b>Course Evaluation Methods</b> Your final grade will be determined by your active participation during the lectures.</p> <p>● <b>Notice for Student</b> This course is an optional subject which does not count towards the number of credits required for graduation in any program at Nagoya University.</p>			
<b>Textbook</b>	---		
<b>Reference Book</b>	---		

<b>Calculus I</b>			
<b>Registration code</b>	0064511	<b>Credits</b>	2.0
<b>Course Category</b>	Sciences Basic	<b>Classroom</b>	C14
<b>Term(Semester)/Day/Period</b>	I (First year, First semester) / Thu / 5 (16:30~18:00)		
<b>Instructor</b>	RICHARD Serge Charles		
<b>Contact</b>	Office: Room 237, Science Building A Email: richard@math.nagoya-u.ac.jp Web site : <a href="http://www.math.nagoya-u.ac.jp/~richard/fall2016.html">http://www.math.nagoya-u.ac.jp/~richard/fall2016.html</a>		
<b>Target Schools (Programs)</b>	Le(J) · La(S) · Ec(S) · Sc(P · C · B) · En(P · C · Au) · Ag(B)		
<p>● <b>Aim of the course</b> Analysis is the field of mathematics that describes and analyzes quantitative changes, and the central methods are differential and integral calculus. These methods are essential techniques in natural science, and have recently found increasing applications also in social sciences. The aim of the first half of this one-year course is to provide a solid understanding of functions of a single variable.</p> <p>● <b>Course Prerequisites</b> Some basic knowledge on calculus from high school is assumed, including differentiation and integration of polynomial functions.</p> <p>● <b>Course Content</b></p> <ol style="list-style-type: none"> <li>1. Limits and continuity Basic properties of limits of sequences and functions, continuous functions and their basic properties, maxima and minima, asymptotic properties of functions.</li> <li>2. Differentiation Basic properties of the derivative and its interpretation, mean value theorem, higher derivatives, Taylor series.</li> <li>3. Integration Riemann integral and its properties, improper integrals, the fundamental theorem of calculus.</li> </ol> <p>● <b>Course Evaluation Methods</b> The final grade will be determined by quizzes (30%), the midterm (30%) and a final exam (40%).</p> <p>● <b>Notice for Students :</b> This course uses the course withdrawal system. To withdraw from the course and obtain the grade Absent the student must submit a written Course Withdrawal Request before the end of November. After that time any student who participated in any part of the examination will be graded S, A, B, C or F.</p>			
<b>Textbook</b>	--		
<b>Reference Book</b>	--		